

# Microstructure and Flavor Release; some novel approaches

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Frozen Dessert Center Conference

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A collage of images related to ice cream. On the left, several tubs of ice cream are shown, including Planet Oat Blueberry Oat Crumble, Wildgood Caramelized Fig, So Delicious Black Tea and Vanilla, Ben &amp; Jerry's Praline Fudge, and So Delicious Chocolate Cookies. In the center, a hand holds a sandwich made with ice cream, cookies, and Oreos. On the right, a graphic titled 'LOW CALORIE ICE CREAMS' lists '290' and '17' and includes images of Halo Top, Høgen-Dazs Heaven, Yasso, and So Delicious Sandwiches. Below the collage is a grey rectangular area with the text 'Relevance?'.

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## Relevance?



<https://www.taste-institute.com/en/resources/blog/importance-of-taste-in-product-development>

3

## Relevance?

*The release of flavor compounds from food, and their delivery to the receptors located in the mouth and nose (Fig. 1), is acknowledged as one of the **key factors determining the perceived flavor quality** of foods.*

Salles et al, 2011

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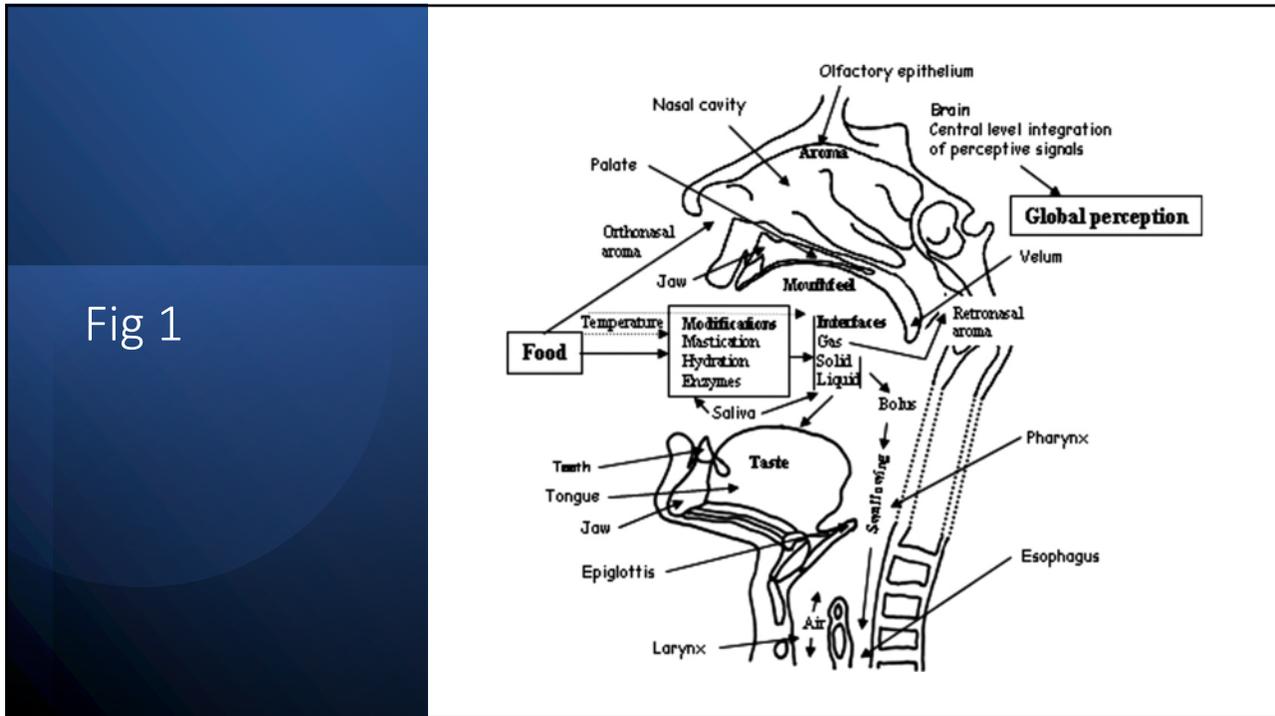
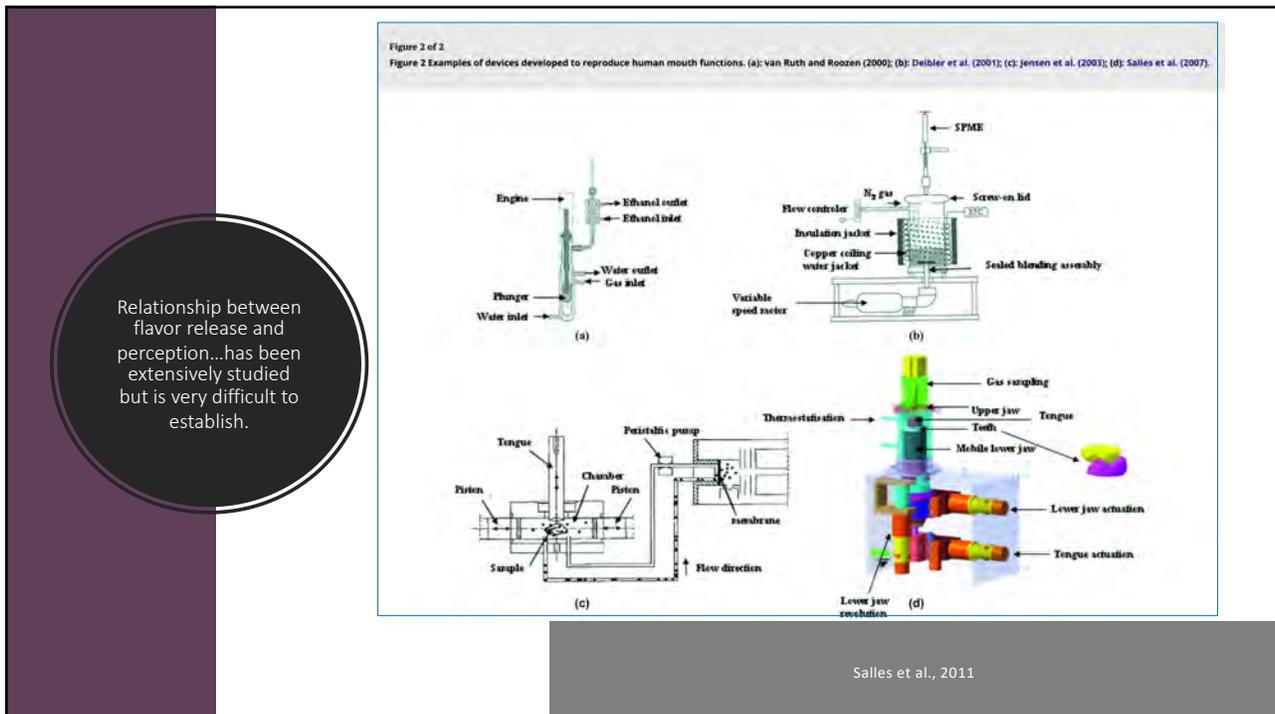


Fig 1

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Relationship between flavor release and perception...has been extensively studied but is very difficult to establish.

Salles et al., 2011

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## Flavor defined

*The flavor perception we derive from eating a food product is determined by the **nature and quantity** of the **flavor components**, the **availability** of these components to the sensory system as a function of time, and the mechanisms and **strategies of perception** and scaling which determine the flavor quality and intensity over time ....*

Overbosch et al., 1991

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## Strong sense of convention

### Flavor

- Vanilla
  - Chocolate
  - Strawberry
  - Dairy
- 
- Texture/mouthfeel
  - Melt
  - Composition
  - Color
  - Etc.



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A few key  
points

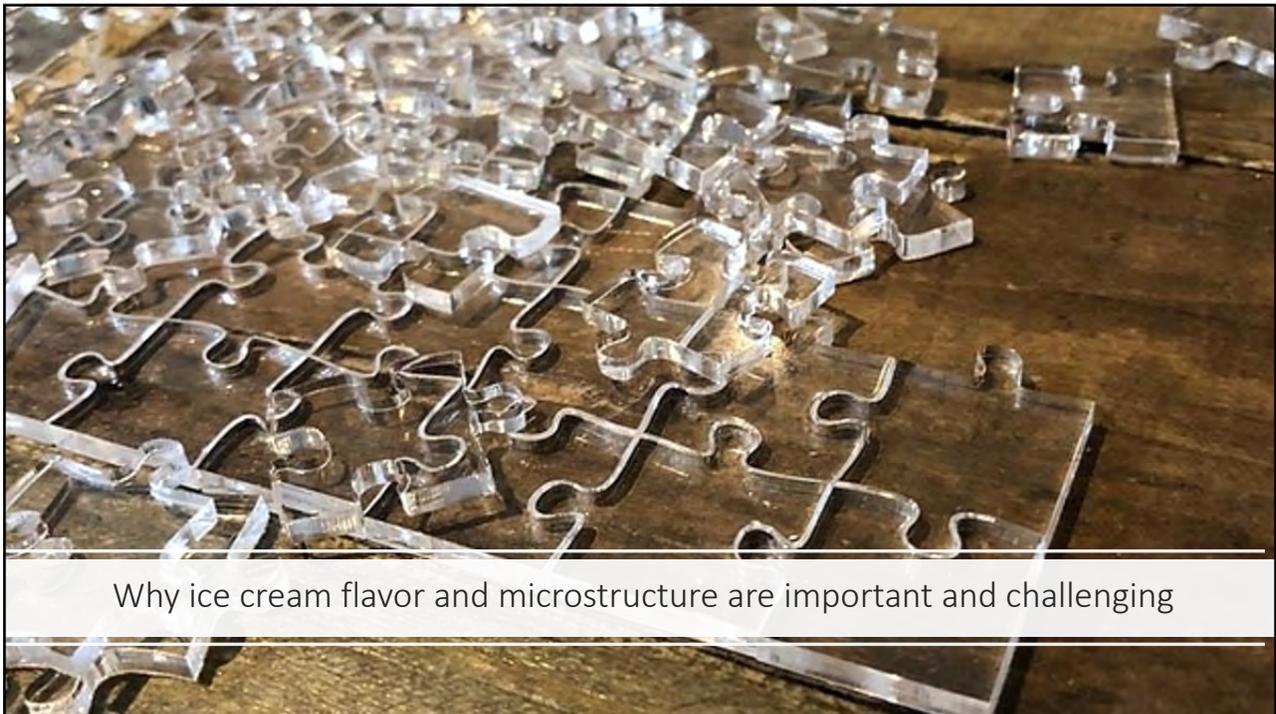
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IC structure is  
complex, so are flavors  
– this is trouble

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Some practical  
principles → novel  
approaches

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Why ice cream flavor and microstructure are important and challenging

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# Complex Ingredients

## Ice Cream And Frozen Dairy Dessert Application Monograph, USDEC

Nutrition Facts	
Serving Size 1/2 Cup (69g) Servings Per Container 16	
Amount Per Serving	
<b>Calories 140</b>	<b>Calories from Fat 60</b>
<b>% Daily Value*</b>	
<b>Total Fat 7g</b>	<b>11%</b>
Saturated Fat 4.5g	23%
Trans Fat 0g	
Cholesterol 30mg	10%
Sodium 65mg	3%
<b>Total Carbohydrate 17g</b>	<b>6%</b>
Dietary Fiber 0g	0%
Sugars 14g	
<b>Protein 2g</b>	
Vitamin A 2%	Vitamin C 0%
Calcium 6%	Iron 0%

\*Percent Daily Values are based on a diet of 2,000 calories. Your daily values may be higher or lower depending on your calorie needs.

	Calories	2,000	2,500
Total Fat	Less than 50g	60g	
Saturated Fat	Less than 20g	25g	
Cholesterol	Less than 300mg	300mg	
Sodium	Less than 2,400mg	2,400mg	
Total Carbohydrate	300g	370g	
Dietary Fiber	25g	30g	

Calories per gram:  
Fat 9 • Carbohydrate 4 • Protein 4

INGREDIENTS: MILK, CREAM, SUGAR, CORN SYRUP, DESSERT SOLIDS (WHEY, WHEY SOLIDS, NONFAT DRY MILK, AND EDIBLE SALTS (SODIUM CARBONATE, SODIUM CITRATE, DIPOTASSIUM PHOSPHATE)), STABILIZER (MONO & DIGLYCERIDES, CELLULOSE GUM, GUAR GUM, LOCUST BEAN GUM, POLYSORBATE 80, CARRAGEENAN, DISODIUM PHOSPHATE, CALCIUM SULFATE), NATURAL AND/OR ARTIFICIAL FLAVORS AND ANNATTO.

U.S. DAIRY INGREDIENTS: COMPOSITIONS AND ADVANTAGES IN FROZEN DESSERTS						
INGREDIENT	PROTEIN (%)	LACTOSE (%)	FAT (%)	ASH (%)	MOISTURE (%)	APPLICATION ADVANTAGE
Skim Milk Powder	34 to 37	49 to 52	0 to 1	8 to 9	3 to 4	Stable source of dairy solids; cost-effective
Evaporated, Condensed Skim Milk	7	11	0	1	80	Cost-effective source of dairy solids
Milk Protein Concentrate	42 to 85	8 to 50	1 to 2	8 to 10	< 5	Common source for milk protein fortification
Milk Protein Isolate	> 90	0.5	< 2	< 8	5	Source of highly concentrated milk protein
Casein*	80 to 85	0.5	1	4 to 6	5	Concentrated casein protein with various functional properties
Sweet Whey Powder	10	> 70	1	9	5	Cost-effective source of milk solids
Whey Protein Concentrate	34 to 80	10 to 50	4 to 6	8 to 7	4	Highly functional, nutritional protein
Whey Protein Isolate	> 90	< 1	< 1	3	4	Concentrated source of highly functional, nutritional protein
Whey Permeate	2 to 6	70 to 85	< 1	10	4	Cost-effective source of milk solids
Lactose	< 1	95+	< 0.1	< 0.5	5	Source of milk sugar. Contributes total solids and freezing point
Whey Protein Phospholipid Concentrate	> 50	1 to 5	> 12	< 8	< 6	Source of protein and milk phospholipid for emulsification and foaming

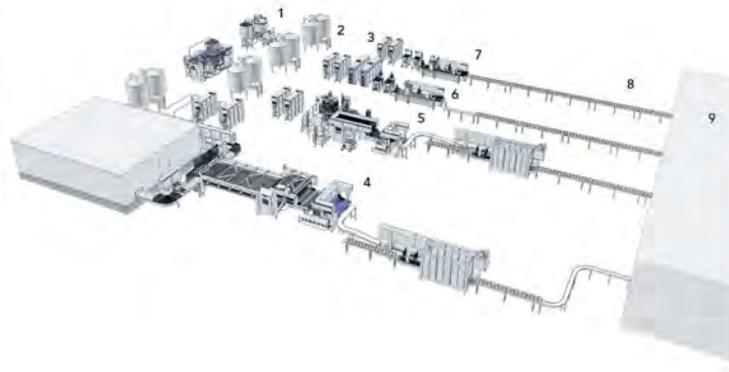
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# Complex processing

Fig.19.10

Ice cream plant for production of 5,000-10,000 l/h of various types of ice cream.

1. Mix preparation
2. Ageing tanks
3. Continuous freezers
4. Bulk filling
5. Cone filling
6. Moulding
7. Extrusion
8. Cartoning
9. Storage



TetraPak, 2023

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# Complex Material

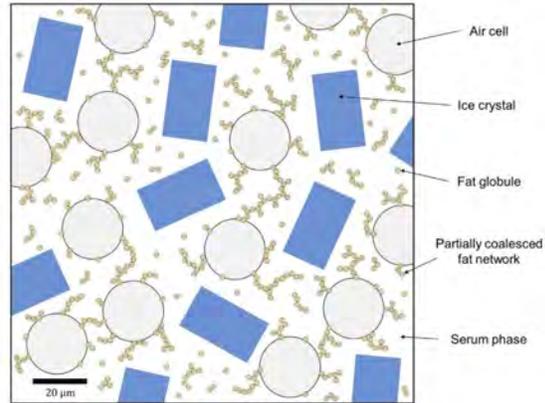


FIGURE 2 Schematic illustration of typical ice cream microstructure

VanWees et al., 2021

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Variation in finished products - survey

United States commercial ice cream products . . .

Table 2-Compositional and structural attributes of commercial ice cream samples.

Components	Mean and standard error	Range
Mean ice crystal size ( $\mu\text{m}$ )	48.1 $\pm$ 2.3	26.3-67.1
Mean air cell size ( $\mu\text{m}$ )	29.9 $\pm$ 1.5	17.1-39.5
Percent total fat (%)	8.6 $\pm$ 1.0	0.01-14.3
Percent fat destabilization (%)	21.9 $\pm$ 3.88	2.60-55.3
Overrun (%)	75.0 $\pm$ 6.5	21.7-119
Density of ice cream (g/L)	649 $\pm$ 03	509-904
Density of ice cream mix (kg/L)	1.10 $\pm$ .01	1.07-1.16
Drip-through rate (g/min)	1.07 $\pm$ 0.15	0.13-1.88
Total solids (%)	38.3 $\pm$ 0.67	31.1-42.6

Table 3-Mean values of compositional and structural analyses and the corresponding Student's *t*-test of significant differences at *P* < 0.05 for the 18 samples analyzed.

Sample code	Mean ice crystal size ( $\mu\text{m}$ )	Mean air cell size ( $\mu\text{m}$ )	Total fat (%)	Fat destabilization (%)	Overrun (%)	Ice cream mix density (kg/L)*	Drip-through rate (g/min)	Total Solids (%)
106	46.3 $\pm$ 5.0 <sup>de</sup>	30.4 $\pm$ 2.0 <sup>bcde</sup>	10.3 $\pm$ 0.1 <sup>def</sup>	15.1 $\pm$ 1.7 <sup>bh</sup>	95.1 $\pm$ 1.1 <sup>bc</sup>	1.10 <sup>ef</sup>	1.20 $\pm$ 0.14 <sup>d</sup>	40.2 $\pm$ 0.6 <sup>bc</sup>
159	26.3 $\pm$ 1.6 <sup>b</sup>	22.2 $\pm$ 1.1 <sup>fbh</sup>	0.1 $\pm$ 0.1 <sup>ij</sup>	11.0 $\pm$ 4.1 <sup>hi</sup>	96.0 $\pm$ 2.9 <sup>b</sup>	1.16 <sup>c</sup>	1.85 $\pm$ 0.11 <sup>a</sup>	36.9 $\pm$ 0.2 <sup>bc</sup>
171	53.0 $\pm$ 2.3 <sup>b</sup>	30.1 $\pm$ 4.1 <sup>cde</sup>	13.1 $\pm$ 0.3 <sup>b</sup>	20.4 $\pm$ 1.6 <sup>g</sup>	67.0 $\pm$ 2.4 <sup>e</sup>	1.09 <sup>g</sup>	0.33 $\pm$ 0.05 <sup>gh</sup>	39.1 $\pm$ 0.3 <sup>cd</sup>
215	53.5 $\pm$ 1.3 <sup>b</sup>	24.6 $\pm$ 0.5 <sup>hfg</sup>	9.0 $\pm$ 0.5 <sup>g</sup>	7.8 $\pm$ 1.0 <sup>jk</sup>	91.8 $\pm$ 2.6 <sup>cd</sup>	1.09 <sup>g</sup>	0.67 $\pm$ 0.06 <sup>e</sup>	36.4 $\pm$ 0.1 <sup>g</sup>
286	40.0 $\pm$ 5.1 <sup>de</sup>	28.0 $\pm$ 5.2 <sup>defg</sup>	0.2 $\pm$ 0.1 <sup>ij</sup>	26.0 $\pm$ 2.7 <sup>ef</sup>	68.4 $\pm$ 2.7 <sup>g</sup>	1.16 <sup>c</sup>	1.72 $\pm$ 0.13 <sup>ab</sup>	34.6 $\pm$ 0.1 <sup>h</sup>
293	50.8 $\pm$ 1.4 <sup>bcd</sup>	26.4 $\pm$ 1.3 <sup>efg</sup>	9.4 $\pm$ 0.4 <sup>fg</sup>	55.3 $\pm$ 5.9 <sup>a</sup>	83.4 $\pm$ 4.3 <sup>ef</sup>	1.13 <sup>d</sup>	0.24 $\pm$ 0.04 <sup>gh</sup>	42.6 $\pm$ 3.2 <sup>a</sup>
313	61.9 $\pm$ 4.9 <sup>a</sup>	27.1 $\pm$ 2.8 <sup>efg</sup>	14.3 $\pm$ 0.6 <sup>a</sup>	6.0 $\pm$ 1.6 <sup>kl</sup>	26.9 $\pm$ 1.1 <sup>i</sup>	1.09 <sup>g</sup>	1.40 $\pm$ 0.32 <sup>cd</sup>	40.9 $\pm$ 0.4 <sup>b</sup>
423	51.7 $\pm$ 1.9 <sup>bc</sup>	36.5 $\pm$ 1.7 <sup>bc</sup>	7.6 $\pm$ 1.0 <sup>h</sup>	2.6 $\pm$ 0.2 <sup>k</sup>	51.0 $\pm$ 1.5 <sup>b</sup>	1.09 <sup>g</sup>	1.69 $\pm$ 0.08 <sup>bd</sup>	35.9 $\pm$ 0.2 <sup>bc</sup>
472	41.3 $\pm$ 1.1 <sup>efg</sup>	38.2 $\pm$ 1.7 <sup>a</sup>	5.7 $\pm$ 0.4 <sup>i</sup>	30.6 $\pm$ 1.0 <sup>de</sup>	119 $\pm$ 4.0 <sup>a</sup>	1.12 <sup>c</sup>	1.26 $\pm$ 0.09 <sup>d</sup>	35.8 $\pm$ 0.3 <sup>bc</sup>
559	42.8 $\pm$ 0.6 <sup>efg</sup>	37.4 $\pm$ 3.7 <sup>a</sup>	*11.5 <sup>c</sup>	45.0 $\pm$ 3.4 <sup>bc</sup>	21.7 $\pm$ 1.1 <sup>j</sup>	1.10 <sup>e</sup>	0.20 $\pm$ 0.20 <sup>gh</sup>	40.7 $\pm$ 0.6 <sup>b</sup>
603	53.6 $\pm$ 1.2 <sup>b</sup>	26.0 $\pm$ 1.2 <sup>efg</sup>	11.0 $\pm$ 0.1 <sup>cd</sup>	21.9 $\pm$ 7.5 <sup>f</sup>	80.3 $\pm$ 1.9 <sup>f</sup>	1.11 <sup>df</sup>	0.26 $\pm$ 0.03 <sup>gh</sup>	38.6 $\pm$ 0.2 <sup>bc</sup>
638	40.9 $\pm$ 2.6 <sup>efg</sup>	36.6 $\pm$ 2.3 <sup>ab</sup>	9.9 $\pm$ 0.2 <sup>efg</sup>	5.0 $\pm$ 0.8 <sup>k</sup>	97.5 $\pm$ 1.7 <sup>b</sup>	1.11 <sup>df</sup>	1.72 $\pm$ 0.07 <sup>ab</sup>	40.9 $\pm$ 0.4 <sup>b</sup>
652	45.8 $\pm$ 3.2 <sup>ef</sup>	21.9 $\pm$ 1.8 <sup>hij</sup>	10.8 $\pm$ 1.7 <sup>cde</sup>	7.9 $\pm$ 2.3 <sup>kl</sup>	72.2 $\pm$ 2.2 <sup>e</sup>	1.10 <sup>e</sup>	0.44 $\pm$ 0.05 <sup>gh</sup>	39.2 $\pm$ 0.1 <sup>cd</sup>
727	43.9 $\pm$ 2.6 <sup>ef</sup>	28.5 $\pm$ 1.6 <sup>def</sup>	5.5 $\pm$ 0.2 <sup>i</sup>	32.3 $\pm$ 2.8 <sup>cd</sup>	89.6 $\pm$ 3.2 <sup>d</sup>	1.11 <sup>df</sup>	0.64 $\pm$ 0.07 <sup>ef</sup>	37.8 $\pm$ 0.3 <sup>ef</sup>
824	45.7 $\pm$ 2.9 <sup>ef</sup>	34.2 $\pm$ 4.9 <sup>abcd</sup>	5.9 $\pm$ 0.2 <sup>i</sup>	47.0 $\pm$ 1.7 <sup>b</sup>	95.4 $\pm$ 0.7 <sup>bc</sup>	1.09 <sup>g</sup>	0.89 $\pm$ 0.16 <sup>c</sup>	*38.0 <sup>ef</sup>
880	67.1 $\pm$ 5.9 <sup>a</sup>	39.5 $\pm$ 1.8 <sup>a</sup>	14.0 $\pm$ 0.6 <sup>ab</sup>	40.4 $\pm$ 7.3 <sup>c</sup>	23.7 $\pm$ 0.6 <sup>ij</sup>	1.10 <sup>ef</sup>	0.13 $\pm$ 0.02 <sup>gh</sup>	40.6 $\pm$ 0.2 <sup>b</sup>
913	66.3 $\pm$ 3.1 <sup>a</sup>	30.1 $\pm$ 3.6 <sup>cde</sup>	5.2 $\pm$ 0.6 <sup>i</sup>	10.0 $\pm$ 0.6 <sup>hij</sup>	85.7 $\pm$ 2.1 <sup>e</sup>	1.07 <sup>h</sup>	1.53 $\pm$ 0.07 <sup>bc</sup>	31.1 $\pm$ 0.1 <sup>i</sup>
957	37.9 $\pm$ 2.5 <sup>ef</sup>	17.1 $\pm$ 1.6 <sup>h</sup>	11.5 $\pm$ 0.5 <sup>c</sup>	9.0 $\pm$ 3.6 <sup>i</sup>	83.3 $\pm$ 1.6 <sup>ef</sup>	1.11 <sup>df</sup>	1.86 $\pm$ 0.31 <sup>a</sup>	41.0 $\pm$ 0.3 <sup>b</sup>

\*Denotes significant differences within  $\pm$  0.04.  
<sup>a,b,c,d,e,f,g,h,i,j,k</sup>Means not sharing the same letter are significantly different ( $\alpha$  = 0.05).

Warren and Hartel, 2014

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## United States commercial

**Table 2–Compositional and :**

Components	Range
Mean ice crystal size ( $\mu\text{m}$ )	26.3–67.1
Mean air cell size ( $\mu\text{m}$ )	17.1–39.5
Percent total fat (%)	0.01–14.3
Percent fat destabilization (%)	2.60–55.3
Overrun (%)	21.7–119
Density of ice cream (g/L)	509–904
Density of ice cream mix (kg/L)	1.07–1.16
Drip-through rate (g/min)	0.13–1.88
Total solids (%)	31.1–42.6

Warren and Hartel, 2014

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Ice cream  
complexity II

<https://tarateaspoon.com/24-exciting-ice-cream-flavors-you-can-make-tara-teaspoon/>

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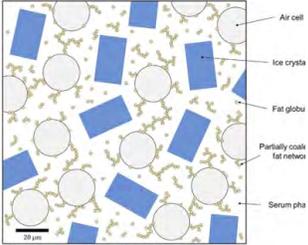


Figure 2 Schematic illustration of typical ice cream microstructure

# During consumption?

*In the case of semi-solids, the main change in food properties is the reduction of viscosity due to shearing forces, temperature change, and dilution or chemical breakdown induced by saliva.*



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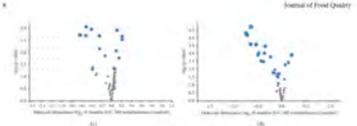


Figure 3 (a-d) Volcano plot of metabolites in 16 samples stored for eight months at 4°C. Each dot within the graph represents a sample analysis. The x-axis represents the log2 fold change of the product abundance of the target metabolite relative to the control (t=0 sample). The y-axis represents the negative log10 of the p-value, representing the concentration of the detected metabolite. Blue dots represent significant differences between the abundance of the metabolite compared with the control, meaning a p-value less than 0.05. The metabolite was identified only if the dot size gives the degree of change (log2 fold change) relative to the control. Thus, an average time treatment from two to eight months, the number of blue metabolites increased.

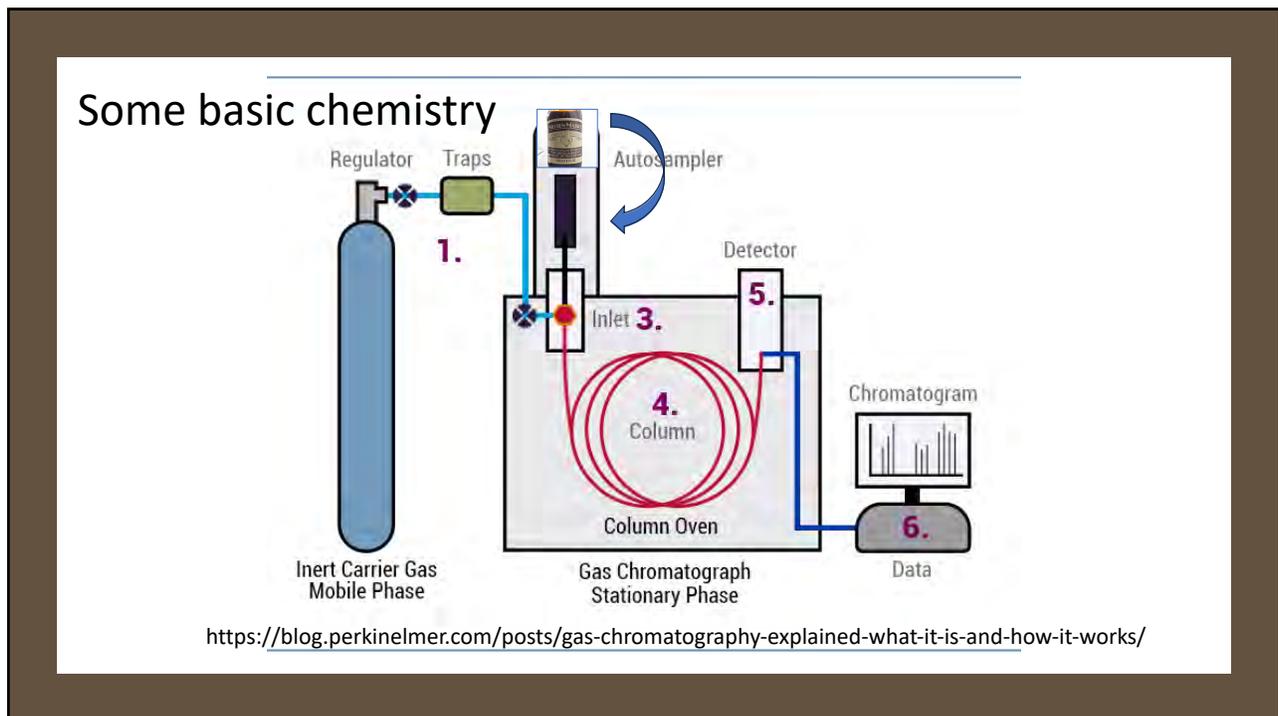
Metabolite	Identification	Reference
Acetic acid	Acetic acid	1
Acetic acid	Acetic acid	2
Acetic acid	Acetic acid	3
Acetic acid	Acetic acid	4
Acetic acid	Acetic acid	5
Acetic acid	Acetic acid	6
Acetic acid	Acetic acid	7
Acetic acid	Acetic acid	8
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Acetic acid	Acetic acid	95
Acetic acid	Acetic acid	96
Acetic acid	Acetic acid	97
Acetic acid	Acetic acid	98
Acetic acid	Acetic acid	99
Acetic acid	Acetic acid	100

# Linking chemical, flavor responses

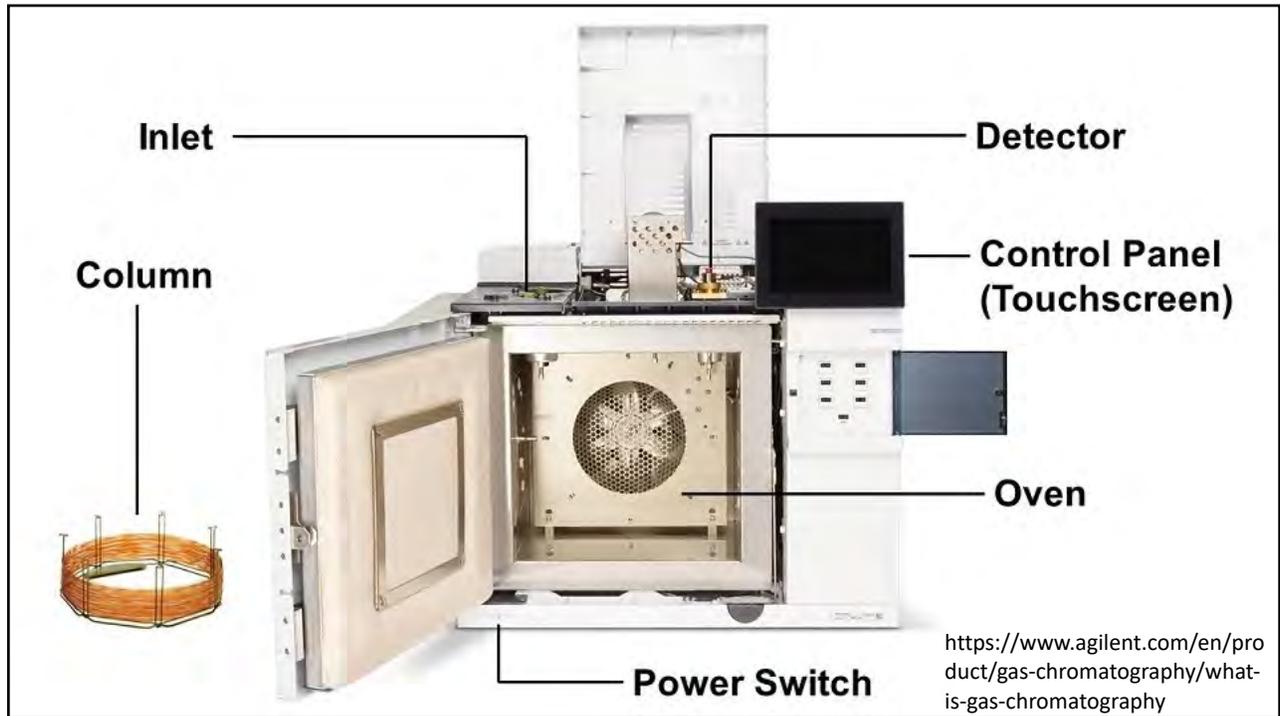
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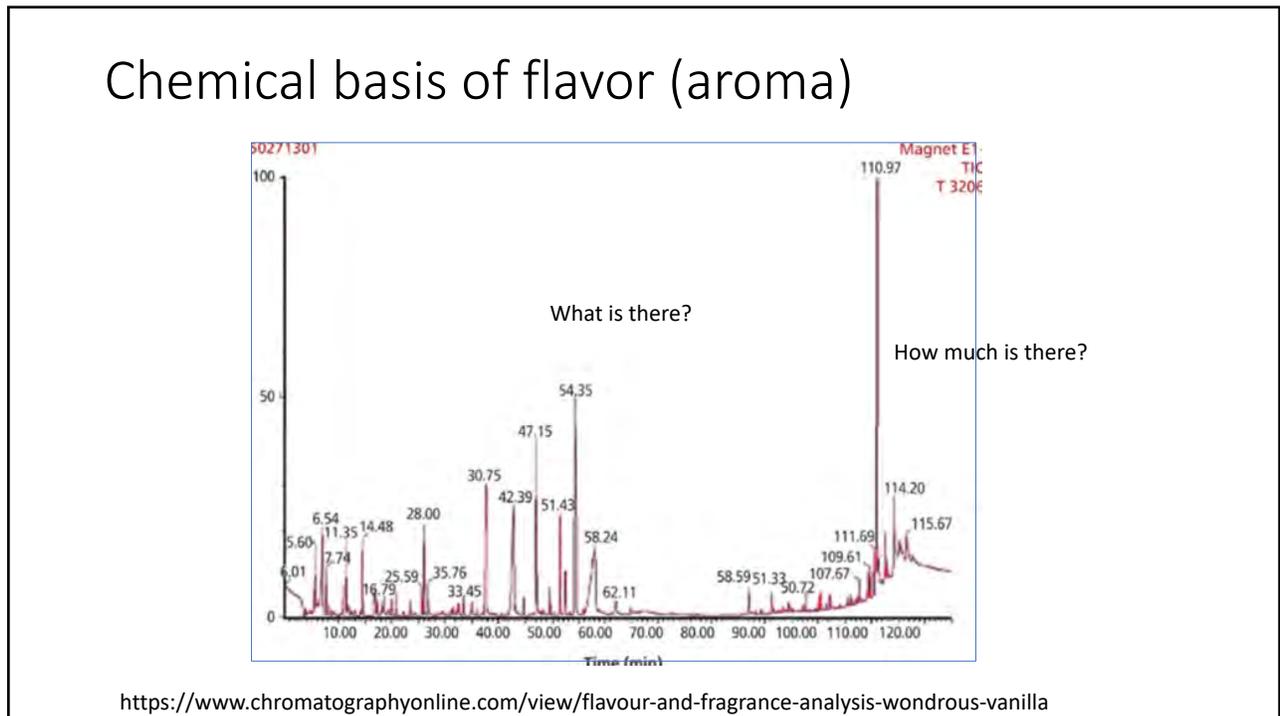
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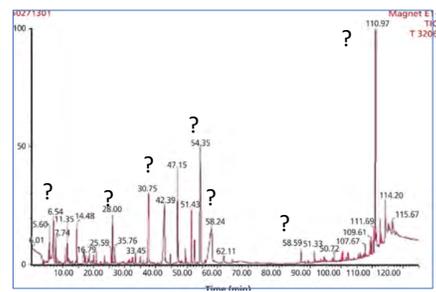
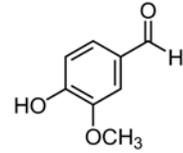
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## What do we know, don't know at this point?

- Compound ID, e.g., 4-hydroxy-3-methoxybenzaldehyde
- Quantity, e.g., 100 mg/kg
- Is there enough to make an *impact*?

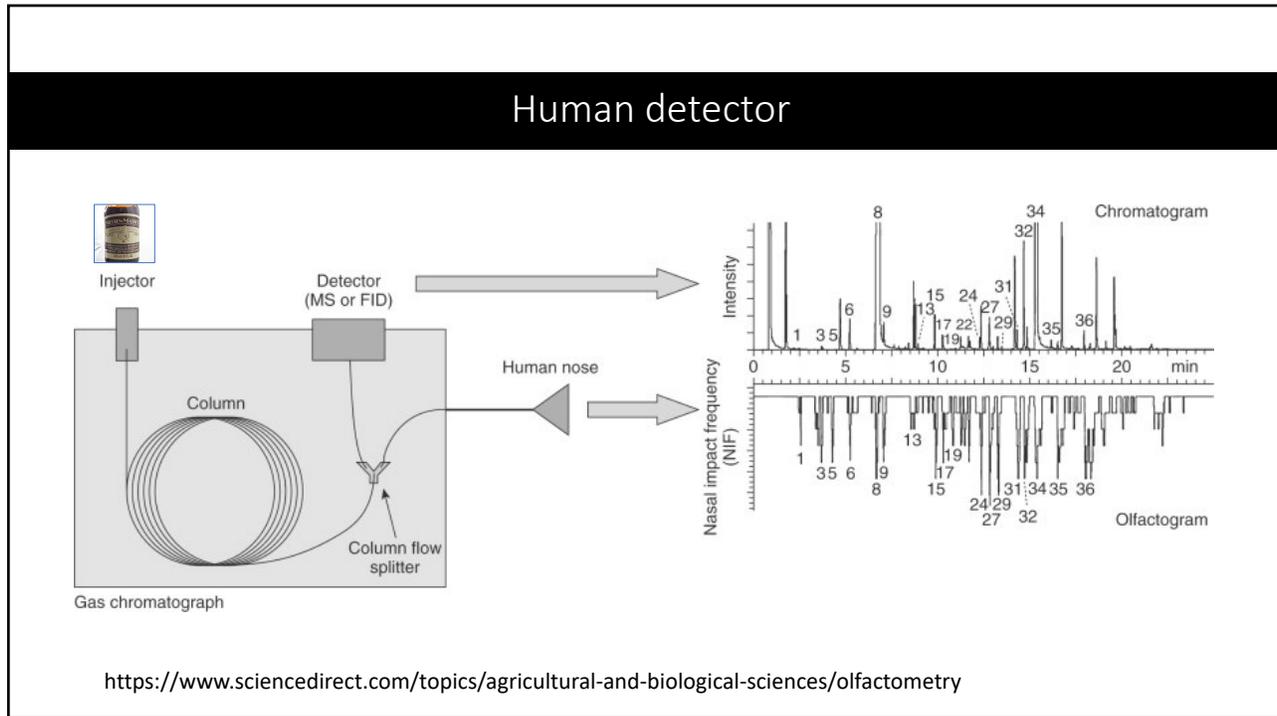


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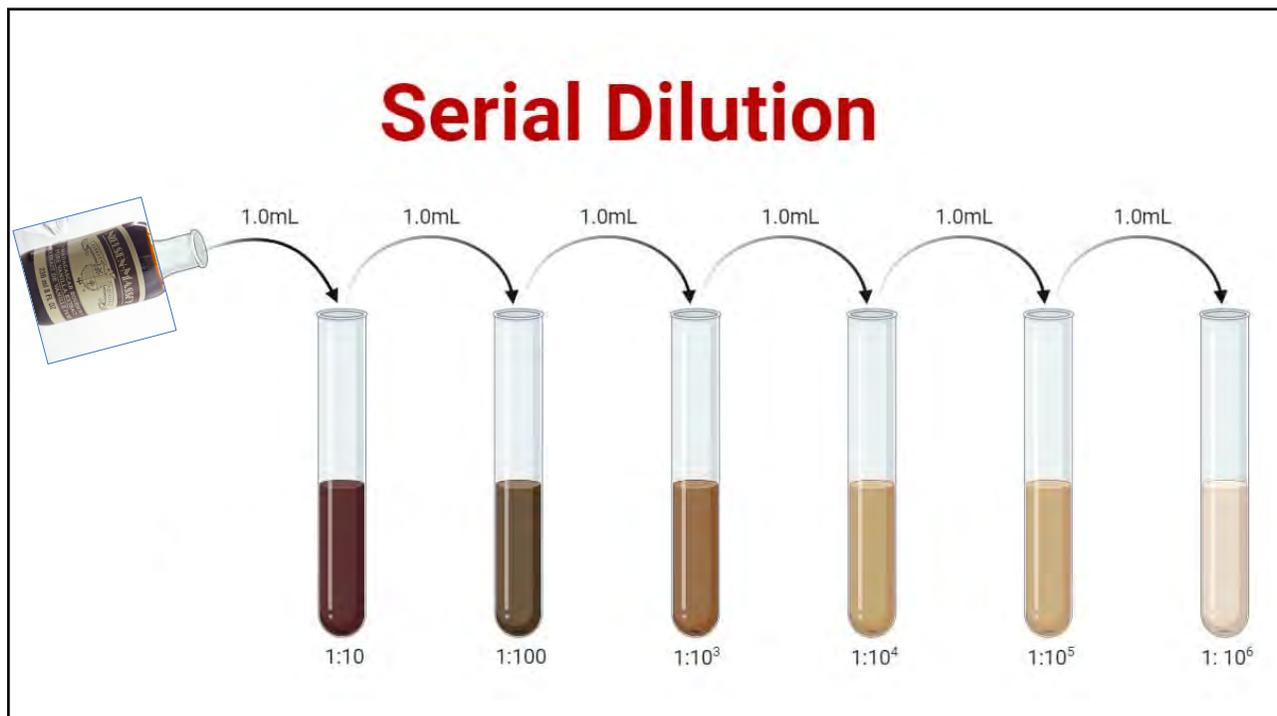


<https://odourobbservatory.org/measuring-odour/gas-chromatography-olfactometry/>

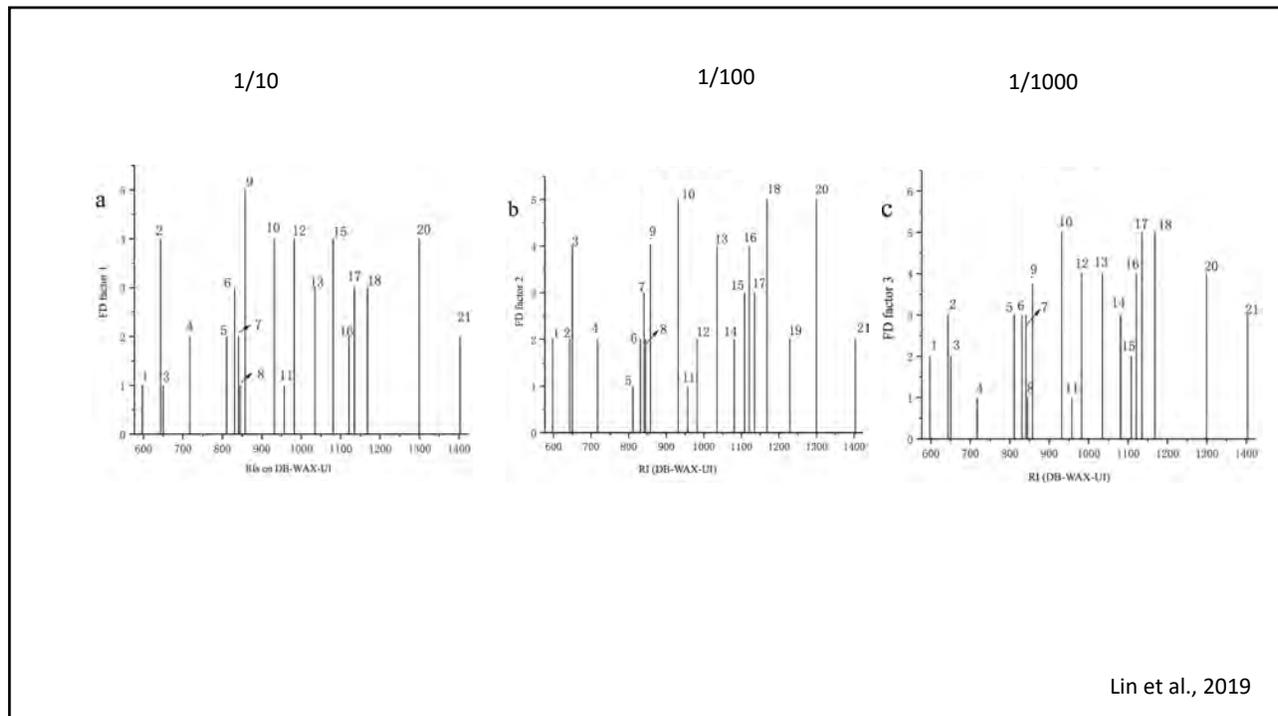
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## Flavor Impact compound: distinct aroma recognition

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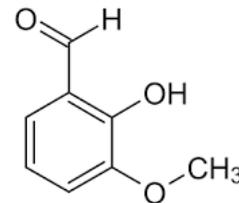
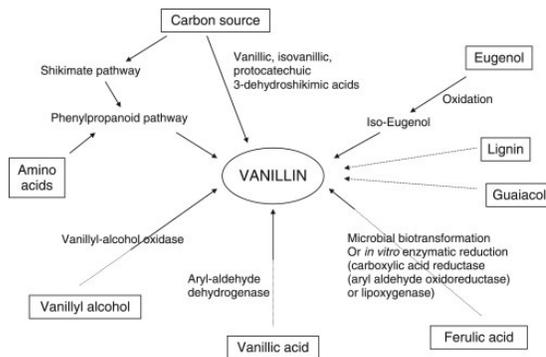
M. TAKAHASHI *et al.***Table 2.** Odor-Active (FD  $\geq$  25) Volatiles in Tahitian Cured Vanilla Beans

Odorant <sup>a</sup>	Odor quality <sup>b</sup>	RI	FD factor	Identification mode <sup>c</sup>
Acetic acid	acidic, sour	1430	625	MS, RI, GC-O
2-Methylbutanoic acid	buttery, cheese-like	1691	25	MS, RI, GC-O
3-Methylbutanoic acid	buttery, cheese-like	1693	25	MS, RI, GC-O
3-Methylnonane-2,4-dione <sup>d</sup>	floral, medicinal	1739	25	RI, GC-O
(2 <i>E</i> ,4 <i>E</i> )-deca-2,4-dienal <sup>d</sup>	oily	1816	25	RI, GC-O
$\beta$ -Damascenone	raisin-like, fruity	1826	25	MS, RI, GC-O
Guaiacol	phenolic, medicinal	1863	125	MS, RI, GC-O
Anisaldehyde	anise-like, raspberry-like	2052	1953125	MS, RI, GC-O
Methyl ( <i>E</i> )-cinnamate	fruity, cinnamon-like	2083	125	MS, RI, GC-O
<i>p</i> -Cresol	fecal	2084	125	MS, RI, GC-O
Anisyl acetate	floral, raisin-like	2132	15625	MS, RI, GC-O
Ethyl ( <i>E</i> )-cinnamate	cinnamon-like, fruity	2145	125	MS, RI, GC-O
Unknown	cooked, meaty	2167	25	GC-O
Eugenol	clove-like, spicy	2169	125	MS, RI, GC-O
4-Vinylguaiacol	phenolic, spicy	2207	25	MS, RI, GC-O
Anisyl alcohol	floral, anise-like	2276	390625	MS, RI, GC-O
Phenylacetic acid	buttery, honey-like	2512	125	MS, RI, GC-O
Vanillin	sweet, vanilla-like	2604	1953125	MS, RI, GC-O
3-Phenylpropanoic acid	metallic, buttery	2672	125	MS, RI, GC-O
Isovanillin	phenolic, medicinal	2718	125	MS, RI, GC-O

28

## Flavor Chemistry reveals

- What chemicals exist
- How much is there (concentration)
- Most impactful compounds (in extract)



29

## Flavor Release

*...is the process whereby flavor molecules move out of a particular molecular environment within a food and into the surrounding saliva or vapor phase (McNulty, 1987; Overbosch et al., 1991)*

*The most important aspect of ...is that aroma molecules leave the bolus and arrive at the olfactory epithelium in the nose where they can be sensed (R. Linforth, A. Taylor, in Flavour in Food, 2006).*

*Flavor release (or lack thereof) is also the basis of some masking agents/technologies that prevent or slow the release of aroma compounds thus decreasing their perception (Rankin, 2023)*



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*Thermodynamic and kinetic parameters control the flavour release from foods. They depend not only on the composition but also on the physical state of the matrix: liquid systems (water or lipids) are relatively simple to study and are useful to understand physical partitioning and release; when moving to viscous or solid systems, more parameters are involved such as diffusion; **real foods are often even more complex, with heterogeneous composition and structure and must be studied at different levels (Voilley and Souchon, 2006)***

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## Mass Transfer

...operations are concerned with the transfer of matter from one stream to another. In many processes a change in phase may also be involved.

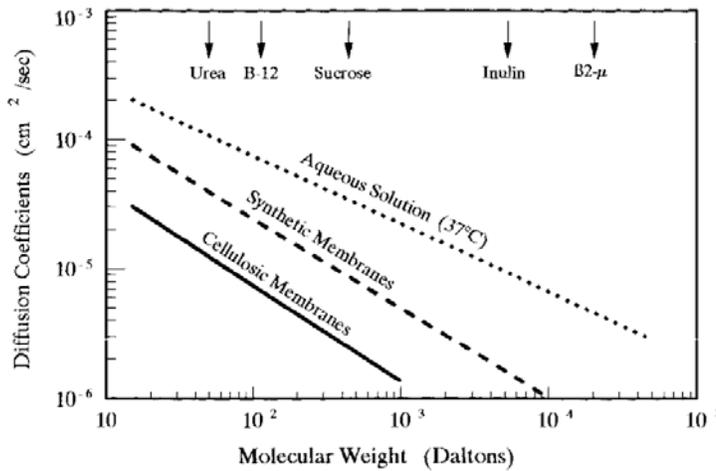
Lewis, 1996

Table 13.1. Some mass transfer process involved in food processing.

Operation	Material transferred	Phase change	Examples and comments
Dehydration	Water	From liquid or solid to vapour	Many drying methods are available. Quality is improved by minimizing the loss of volatile components
Solvent extraction	Oil	From solid matrix to organic liquid	Extraction of oils and fats from animal plant or microbial sources
Leaching	Soluble components	From solid matrix to aqueous solvent	Tea, coffee, sugar extraction; plant protein
Sulphiting	Sulphur dioxide	From gas to solid or liquid media	Sulphiting may also be done using solutions of sodium bisulphite or metabisulphite
Smoking	Phenolic components	From vapour to solid matrix	Preservation of foods by use of antimicrobial agents
Distillation	Alcohol and other volatile components	From liquid to vapour	Recovery of alcohol from a fermentation broth
Packaging	Gases and vapours	From the external environment into the package	Prevention of microbial and oxidation reactions
Membrane processing	Water, and dissolved solutes	From liquid to liquid through a semipermeable membrane	Reverse osmosis for concentrating liquids. Ultrafiltration for concentrating proteins
Oxygen transfer	Oxygen	From gas to liquid	Aerobic fermentation processes in which oxygen is consumed from the solution by micro-organisms
Ion exchange	Metal ions, proteins	From solution to a solid phase	Elution involves the reversal of the adsorption process

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# Diffusivity



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# Effect of lipid type, phase

TABLE 1  
Comparison of Dynamically Released Quantities of Flavor ( $\mu\text{g}$ , after 30 s) from Emulsions Comprising Different Lipid Phases<sup>a</sup>

	Lipids used in emulsions						
	Water <sup>c</sup>	Liquid lipids <sup>b</sup>				Solid lipids <sup>b</sup>	
		Triacetin	Tributyrin	Miglyol <sup>d</sup>	Butter oil <sup>d</sup>	Trimyristin <sup>d</sup>	Tripalmitin
Diacetyl	1.22 <sup>a</sup> ± 0.06	2.02 <sup>b</sup> ± 0.01	2.28 <sup>c</sup> ± 0.14	1.50 <sup>d</sup> ± 0.11	5.95 <sup>e</sup> ± 0.07	2.26 <sup>f</sup> ± 0.17	1.75 <sup>g</sup> ± 0.15
Isobutyl acetate	0.81 <sup>a</sup> ± 0.01	0.70 <sup>b</sup> ± 0.03	0.20 <sup>c</sup> ± 0.00	0.25 <sup>d</sup> ± 0.00	0.33 <sup>e</sup> ± 0.02	0.31 <sup>f</sup> ± 0.01	0.31 <sup>g</sup> ± 0.03
Ethyl 2-methylbutyrate	1.22 <sup>a</sup> ± 0.01	0.91 <sup>b</sup> ± 0.04	0.15 <sup>c</sup> ± 0.01	0.17 <sup>c</sup> ± 0.01	0.25 <sup>d</sup> ± 0.01	0.21 <sup>c,d</sup> ± 0.01	0.35 <sup>e</sup> ± 0.04
(Z)-3-Hexenyl acetate	6.22 <sup>a</sup> ± 0.22	3.25 <sup>b</sup> ± 0.15	0.33 <sup>c</sup> ± 0.02	0.48 <sup>c</sup> ± 0.01	0.69 <sup>d</sup> ± 0.01	0.68 <sup>d</sup> ± 0.02	1.48 <sup>e</sup> ± 0.11
2,3-Dimethylpyrazine	0.28 <sup>a</sup> ± 0.00	0.23 <sup>b</sup> ± 0.03	0.30 <sup>b,c</sup> ± 0.05	0.27 <sup>b,b</sup> ± 0.01	0.31 <sup>b,c</sup> ± 0.00	0.34 <sup>c</sup> ± 0.01	0.24 <sup>b</sup> ± 0.00
(Z)-3-Hexenol	0.74 <sup>a,b</sup> ± 0.01	0.73 <sup>b</sup> ± 0.06	0.58 <sup>c</sup> ± 0.07	0.60 <sup>c</sup> ± 0.00	0.80 <sup>b,d</sup> ± 0.01	0.82 <sup>d</sup> ± 0.01	0.61 <sup>c</sup> ± 0.01
2-Isobutylthiazole	4.50 <sup>a</sup> ± 0.05	2.80 <sup>b</sup> ± 0.21	0.58 <sup>c</sup> ± 0.05	0.67 <sup>c</sup> ± 0.00	0.95 <sup>d</sup> ± 0.02	0.93 <sup>d</sup> ± 0.03	1.93 <sup>e</sup> ± 0.14
Furfuryl acetate	1.23 <sup>a</sup> ± 0.07	0.98 <sup>b</sup> ± 0.10	0.35 <sup>c</sup> ± 0.05	0.52 <sup>d</sup> ± 0.01	0.71 <sup>e</sup> ± 0.03	0.69 <sup>e</sup> ± 0.02	0.79 <sup>e</sup> ± 0.05
Linalool	1.98 <sup>a</sup> ± 0.10	1.13 <sup>b</sup> ± 0.15	0.15 <sup>c</sup> ± 0.02	0.18 <sup>c</sup> ± 0.00	0.26 <sup>c</sup> ± 0.01	0.27 <sup>c</sup> ± 0.01	0.42 <sup>d</sup> ± 0.01
2-Pentylpyridine	2.36 <sup>a</sup> ± 0.14	0.75 <sup>b</sup> ± 0.28	0.21 <sup>c</sup> ± 0.06	0.24 <sup>c</sup> ± 0.09	0.22 <sup>c</sup> ± 0.03	0.19 <sup>c</sup> ± 0.02	0.30 <sup>c</sup> ± 0.04
o-Carvone	1.31 <sup>a</sup> ± 0.01	0.64 <sup>b</sup> ± 0.13	0.11 <sup>c</sup> ± 0.02	0.14 <sup>c</sup> ± 0.01	0.17 <sup>c</sup> ± 0.01	0.18 <sup>c</sup> ± 0.01	0.46 <sup>d</sup> ± 0.04
$\beta$ -Damascenone	4.79 <sup>a</sup> ± 0.43	0.89 <sup>b</sup> ± 0.28	0.20 <sup>c</sup> ± 0.06	0.13 <sup>c</sup> ± 0.04	0.11 <sup>c</sup> ± 0.01	0.13 <sup>c</sup> ± 0.02	0.34 <sup>c</sup> ± 0.04
$\gamma$ -Nonalactone	0.11 <sup>a</sup> ± 0.02	0.07 <sup>b</sup> ± 0.03	0.05 <sup>b</sup> ± 0.01	0.04 <sup>b</sup> ± 0.00	0.05 <sup>b</sup> ± 0.00	0.07 <sup>b</sup> ± 0.01	0.05 <sup>b</sup> ± 0.01
CV <sup>e</sup> (%)	4.2	15.1	13.8	8.3	4.7	6.0	8.0
Log $P^f$		0.36	3.31	10.7 <sup>g</sup>		18.0	20.9
Molarity (mol L <sup>-1</sup> )		0.27	0.17	0.09	0.06	0.06	0.05

<sup>a</sup>Values with different roman letters within a line are significantly different [ANOVA and Duncan's multiple range (DMR) test,  $P < 0.05$ ].

<sup>b</sup>At 22°C.

<sup>c</sup>Adapted from Reference 13.

<sup>d</sup>Adapted from Reference 4.

<sup>e</sup>Average CV.

Rabe et al., 2003

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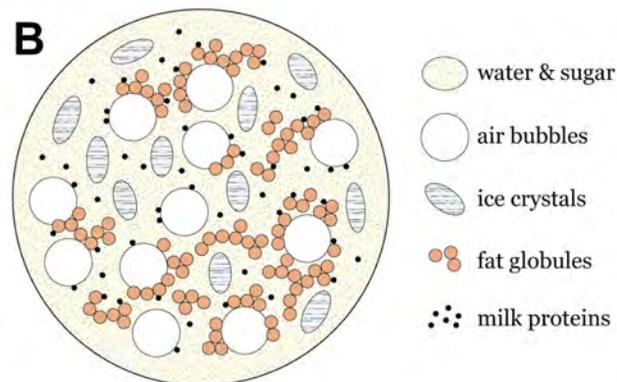
## Effect of lipid type, phase

**TABLE 1**  
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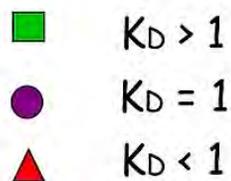
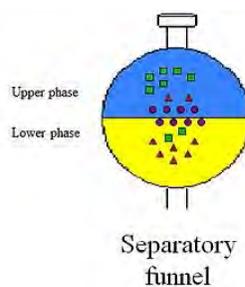
35

## Unraveling flavorant/microstructure physics



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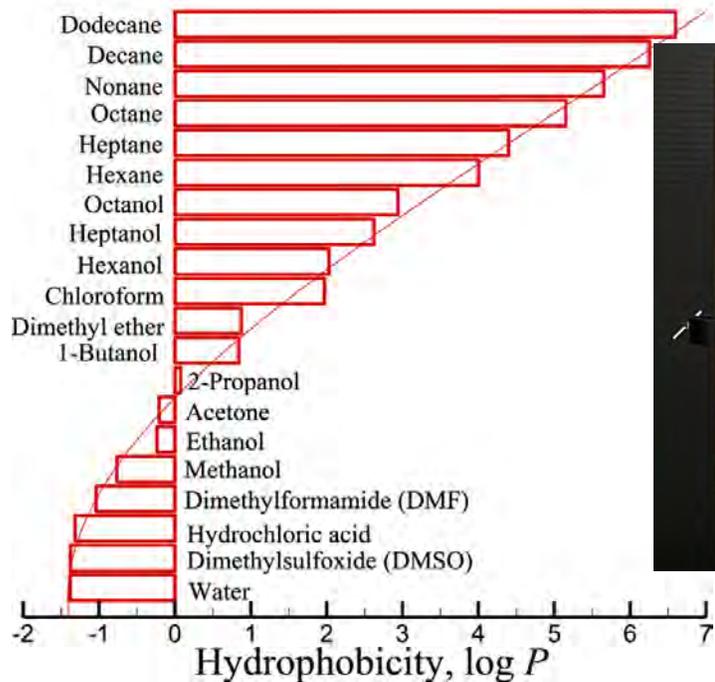
## Partitioning



$$K_D = \frac{[A]_{upper}}{[A]_{lower}}$$

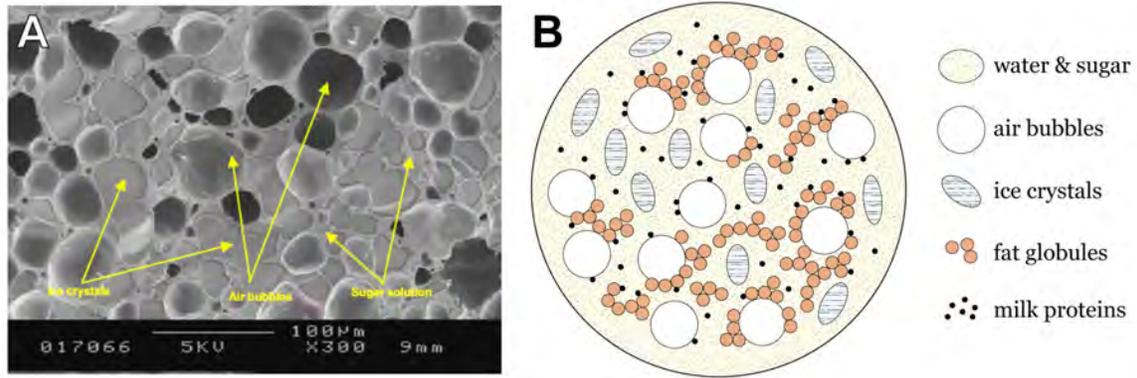


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38

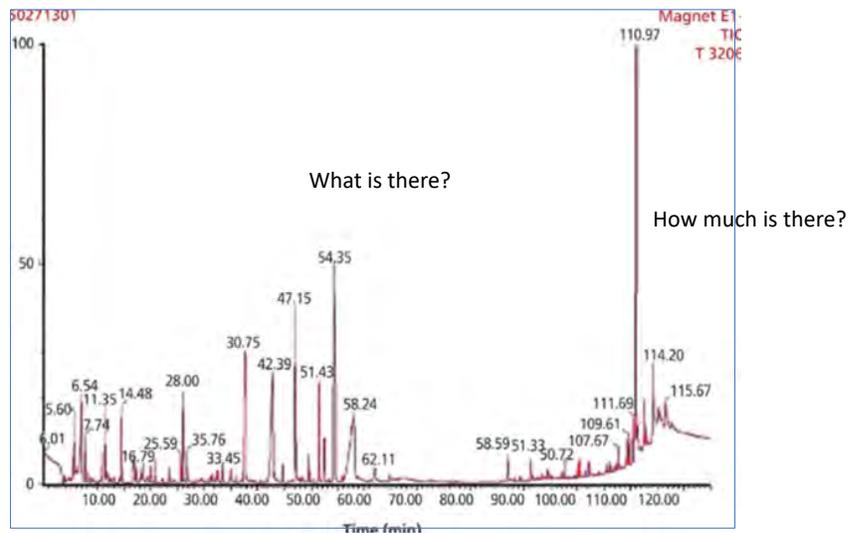
# Matrix/structure effects



<https://scienceandfooducla.wordpress.com/2013/03/05/homemade-ice-cream/>

39

# Chemical basis of flavor (aroma)



<https://www.chromatographyonline.com/view/flavour-and-fragrance-analysis-wondrous-vanilla>

40

# Flavor Impact compound: not so distinct aroma

604

M. TAKAHASHI *et al.***Table 2.** Odor-Active (FD  $\geq$  25) Volatiles in Tahitian Cured Vanilla Beans

Odorant <sup>a</sup>	Odor quality <sup>b</sup>	RI	FD factor	Identification mode <sup>c</sup>
Acetic acid	acidic, sour	1430	625	MS, RI, GC-O
2-Methylbutanoic acid	buttery, cheese-like	1691	25	MS, RI, GC-O
3-Methylbutanoic acid	buttery, cheese-like	1693	25	MS, RI, GC-O
3-Methylnonane-2,4-dione <sup>d</sup>	floral, medicinal	1739	25	RI, GC-O
(2 <i>E</i> ,4 <i>E</i> )-deca-2,4-dienal <sup>d</sup>	oily	1816	25	RI, GC-O
$\beta$ -Damasconone	raisin-like, fruity	1826	25	MS, RI, GC-O
Guaiacol	phenolic, medicinal	1863	125	MS, RI, GC-O
Anisaldehyde	anise-like, raspberry-like	2052	1953125	MS, RI, GC-O
Methyl ( <i>E</i> )-cinnamate	fruity, cinnamon-like	2083	125	MS, RI, GC-O
<i>p</i> -Cresol	fecal	2084	125	MS, RI, GC-O
Anisyl acetate	floral, raisin-like	2132	15625	MS, RI, GC-O
Ethyl ( <i>E</i> )-cinnamate	cinnamon-like, fruity	2145	125	MS, RI, GC-O
Unknown	cooked, meaty	2167	25	GC-O
Eugenol	clove-like, spicy	2169	125	MS, RI, GC-O
4-Vinylguaiaicol	phenolic, spicy	2207	25	MS, RI, GC-O
Anisyl alcohol	floral, anise-like	2276	390625	MS, RI, GC-O
Phenylacetic acid	buttery, honey-like	2512	125	MS, RI, GC-O
Vanillin	sweet, vanilla-like	2604	1953125	MS, RI, GC-O
3-Phenylpropanoic acid	metallic, buttery	2672	125	MS, RI, GC-O
Isovanillin	phenolic, medicinal	2718	125	MS, RI, GC-O

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## 3.2.12 LogP Vanillin

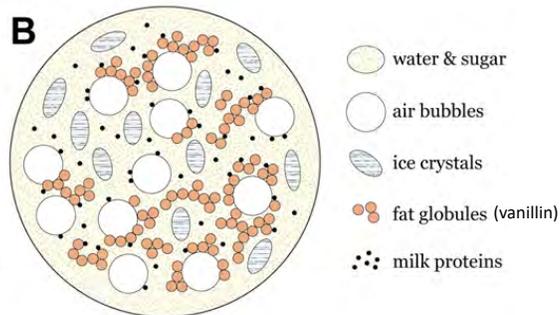
log Kow = 1.37

Hansch, C., Leo, A., D. Hoekman. *Exploring QSAR - Hydrophobic, Electronic, and Steric Constants*. Washington, DC: American Chemical Society, 1995, p. 42

▶ [Hazardous Substances Data Bank \(HSDB\)](#)



25:1

**B**

42



Vanilla: complex chemistry, diffusivity, mass transfer, matrix/composition...

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## Alter matrix to improve flavor....?

*No consistent and complete set of data is available in the literature for comprehensive model validation. In particular, the reported experimental data lack information on the temperature dependence of the diffusion coefficient in the polymer membrane and on the average number of unit cell in the foam layer. This data is critically important for the development of reliable foam diffusion models.*



*Pilon, 2000; Georgia Institute of Technology,*

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## Flavor complexity: consider the music model

The diagram shows a sheet of music for 'Prelude I' by J.S. Bach, Op. 10, No. 1. Three blue arrows point from the sheet music to three different representations of the music: a photograph of a pianist performing on a stage, a pair of modern wooden speakers, and a vintage gramophone with a large horn.

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A few key  
points

---

IC structure is  
complex, so are flavors  
– this is trouble

---

Some practical  
principles → novel  
approaches

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## A few words of caution

- These are new approaches, lack of confidence
- Principles are not universally applicable
- Re-think how we approach food flavor
- Are they really novel?



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## A new approach Illustrated

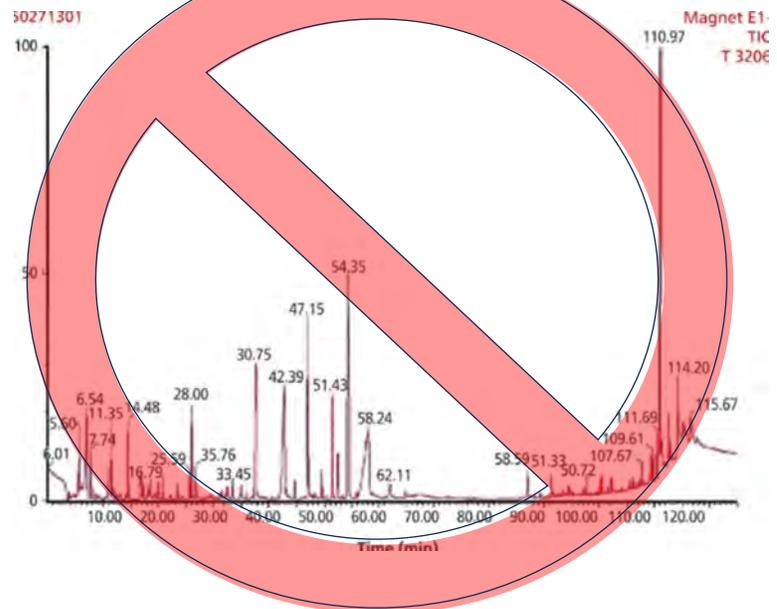


Camilla Arndal Andersen, 2019

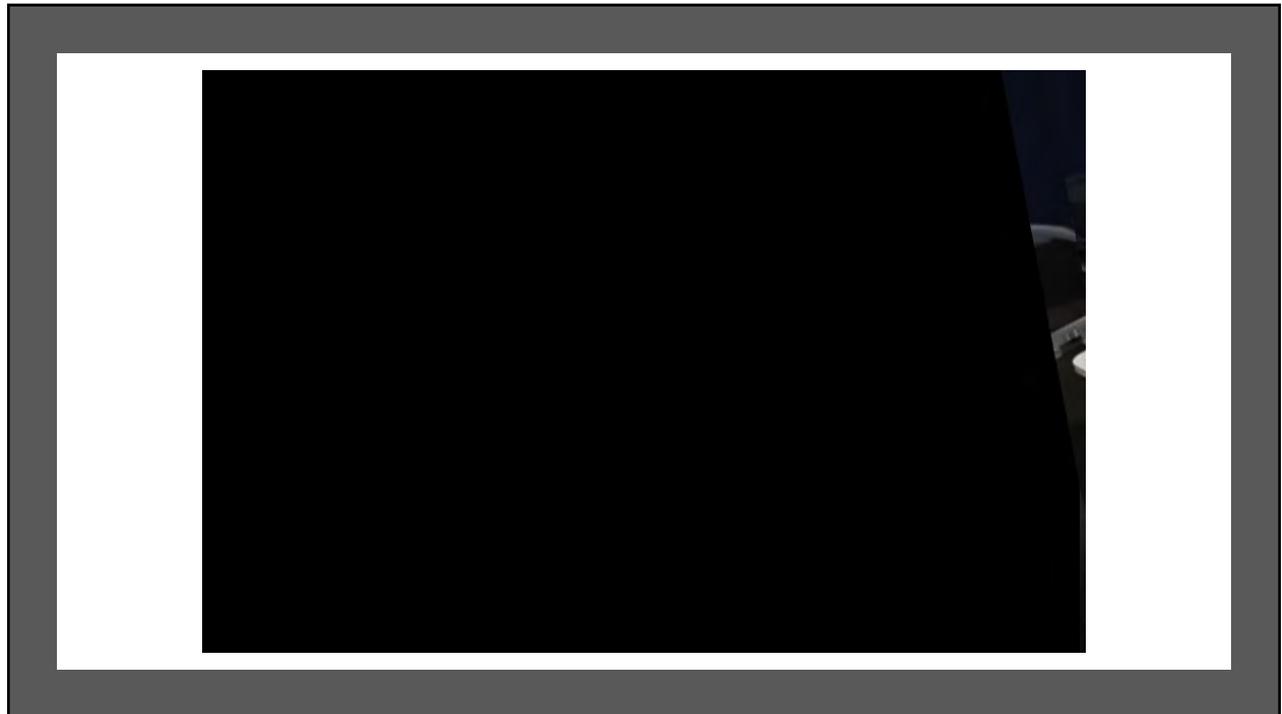
48

# Practical principle #1

Non chemistry approaches:  
Label, color, temperature,  
messaging, etc



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Practical  
principle #2

---

Significant latitude in many flavor, taste systems



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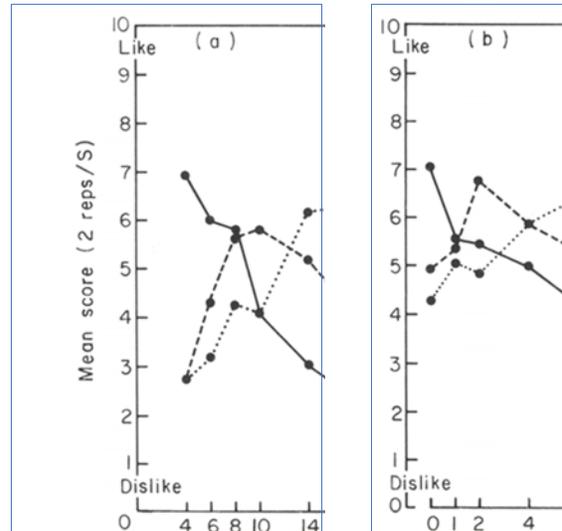
322 R. M. PANGBORN AND M. E. GIOVANNI

**FIGURE 1.** Mean hedonic responses to increasing concentrations of (a) sucrose in lemonade and (b) to fat in milk, subdivided by the subjects' liking for low, medium, or high levels. Each point represents the mean of two replications for the number of subjects represented. ●—●, Low (3, 21); ●---●, medium (38, 26); and ●-----●, high (10, 6) levels. Figures in parentheses are the number of subjects represented for sucrose and butterfat respectively.



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## Relationship: sugar, fat to liking?



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## Implications?

- Sweetness, calories?
- Milkfat, calories?
- Flavor systems



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# Practical principle #3

Flavor preference is fluid



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*"We like what we eat more than we eat what we like" -Kurt Lewin*



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Learned preferences

Bertino et al, 1986

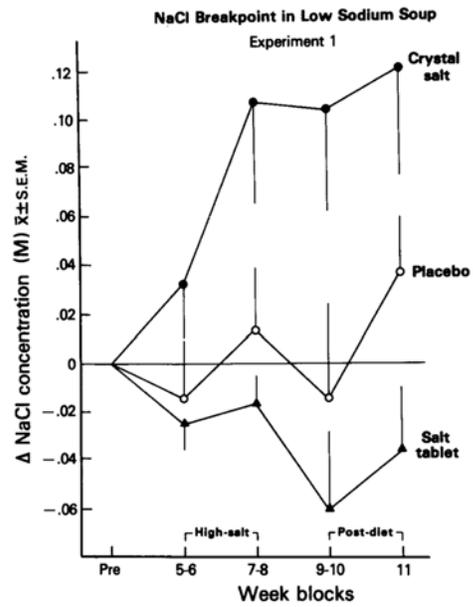


FIG. 1. Change in breakpoint concentration of NaCl in soup during Experiment 1: Crystal salt (n=7); Salt tablet (n=6); Placebo (n=6).

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Implications?

- Keep presence in market



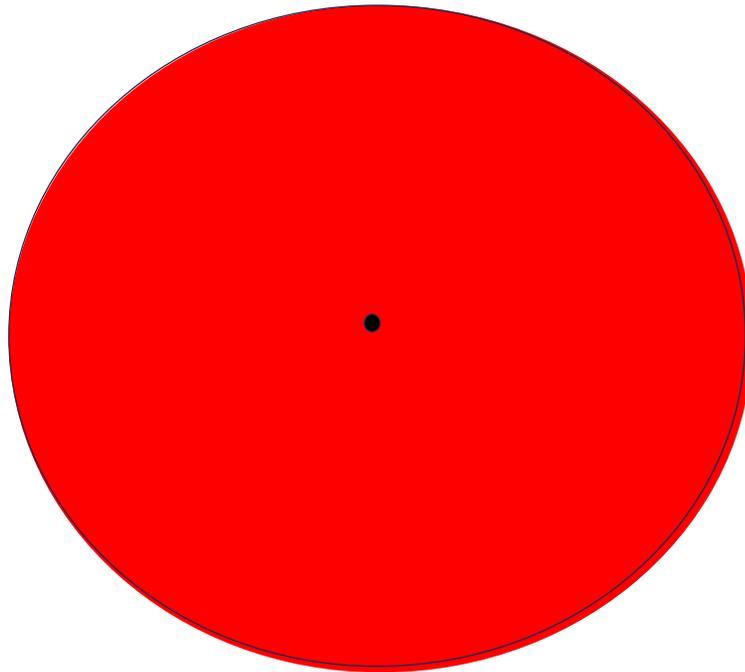
58

# Practical principle #4

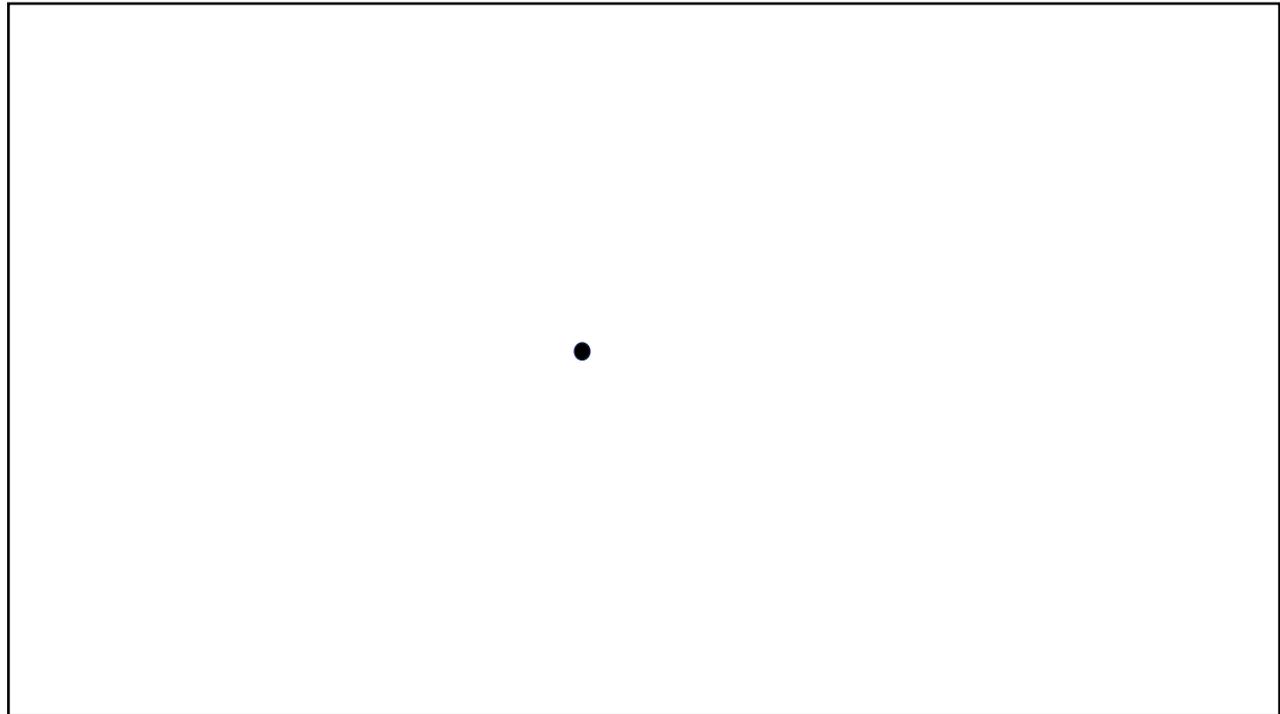
Habituation effects



59

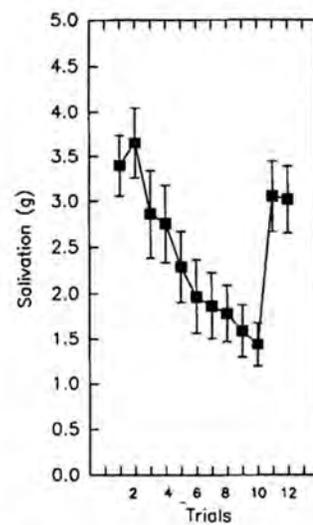


60



61

## Habituation effects

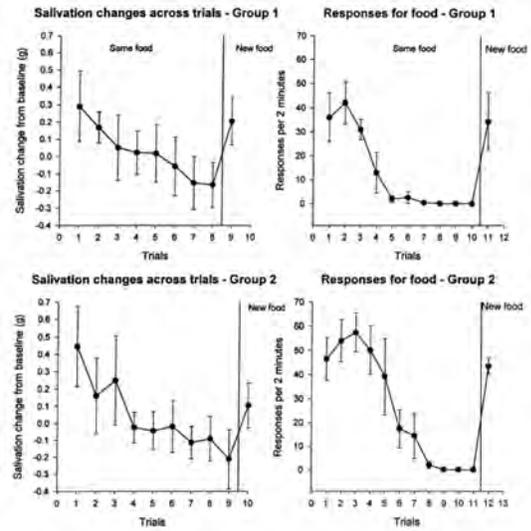


**Figure 1.** Salivation (mean  $\pm$  SEM) for subjects who received lemon or lime juice as the habituating stimulus in trials 1–10, the other juice as the dishabituate in trial 11, and presentation of the habituating stimulus in trial 12. Adapted from (Epstein et al., 1992). Copyright 1992 by Pergamon Press. Reprinted by permission.

Epstein et al., 2009

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# Habituation effects

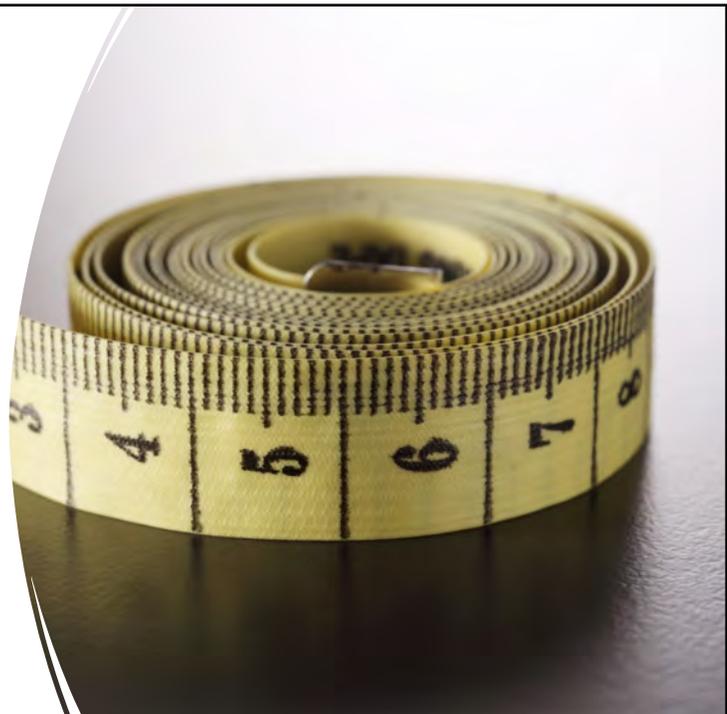


**Figure 3.** Salivation (left graphs, mean  $\pm$  SEM) and motivation (right graphs, mean  $\pm$  SEM) for subjects who were presented cheeseburgers followed by apple pie as the new food. The introduction of the new food was delayed one trial for Group 2 in relationship to Group 1 test whether the recovery of responding occurred after presentation of the new food. Adapted from (Epstein et al., 2003). Copyright 2003 by Elsevier Ltd. Reprinted by permission.

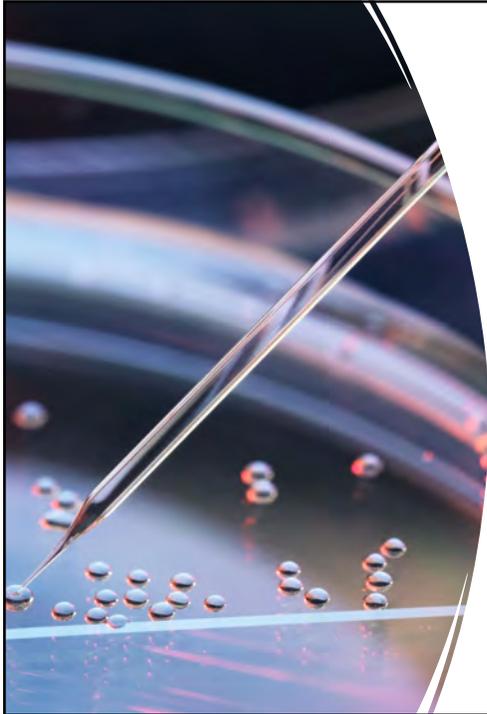
63

# Implications?

- Stay in market
- Incremental changes



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# Microstructure and Flavor Release; some novel approaches

S.A. Rankin, Ph.D.

Oct, 2023

Frozen Dessert Center Conference